

Matthias Renka November 16 - 17, 2019 34. Hofer Vliesstofftage, Hof - Germany





CONTENT



Chemically bonded Nonwovens - Vorteile und Grenzen verschiedener Dispersionsbindemittel



- 1. Synthomer at glance
- 2. General information
- 3. Chemical bonding process
- 4. All about polymerisation
- 5. Different chemistries, different properties
- 6. Summary



SYNTHOMER AT GLANCE

Leading in specialty polymers





Top 5

Global supplier of emulsion and speciality polymers



£ 1.61 Bn
Group revenue



25 Production sites



>10

Polymer classes xSBR, HS-SBR, SA, PA, NBR, CR, VP, VAc, VA-Co, NR



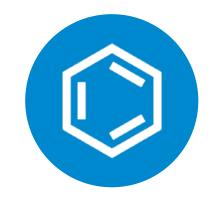
FTSE250

Listed company
Focused on organic
growth and M&A



2900 Employees





>20%

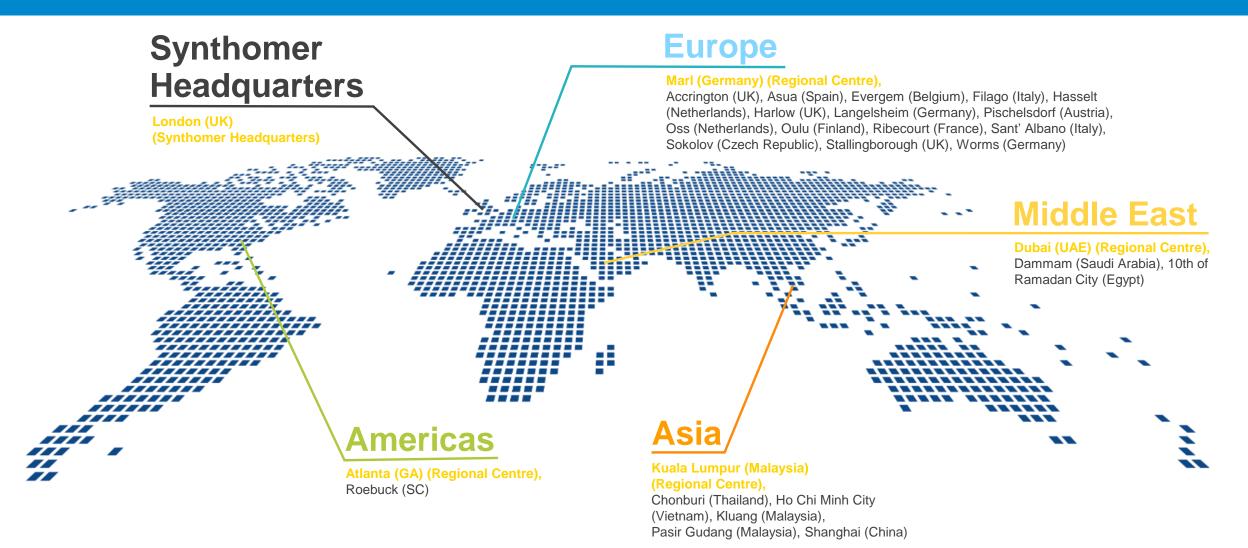
Turnover with products younger then 5 years



SYNTHOMER AT GLANCE

Your global partner







GENERAL INFORMATION

Woven vs. Nonwoven production process



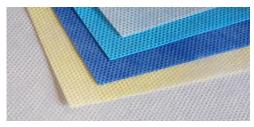
Woven fabric production process



Nonwoven fabric production process



Web forming & bonding



Nonwoven fabric

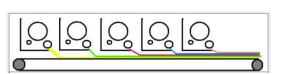


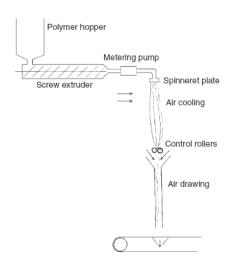
GENERAL INFORMATION

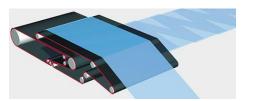
Different methods of web formation

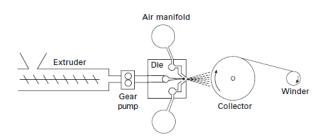


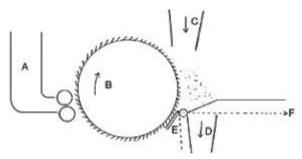
- Parallel Laying
- Cross Laying
- Air Laying
- Spun Laying
- Melt Blown
- Wet Laying
- ...

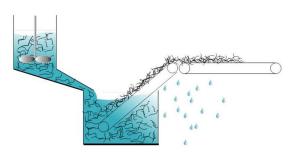










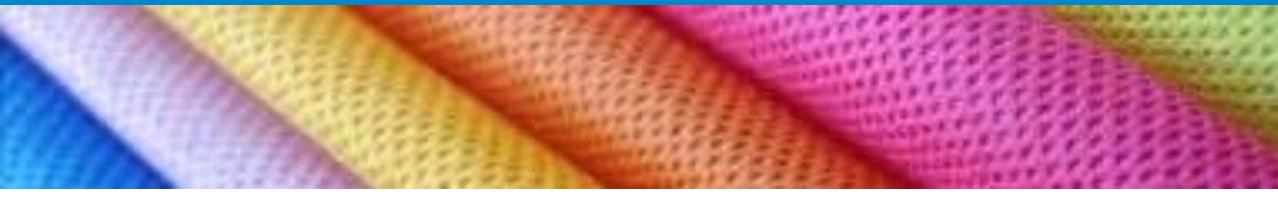




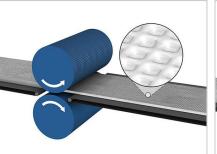
GENERAL INFORMATION

Different methods of web bonding

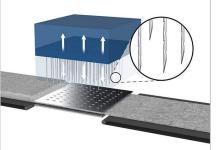


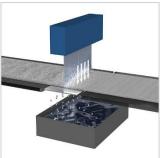


- Thermal bonding
- Needle punching
- Hydro entanglement
- Stitch bonding
- Solvent bonding
- Chemical bonding













Why do we need chemical bonding?

Improve characteristics such as:

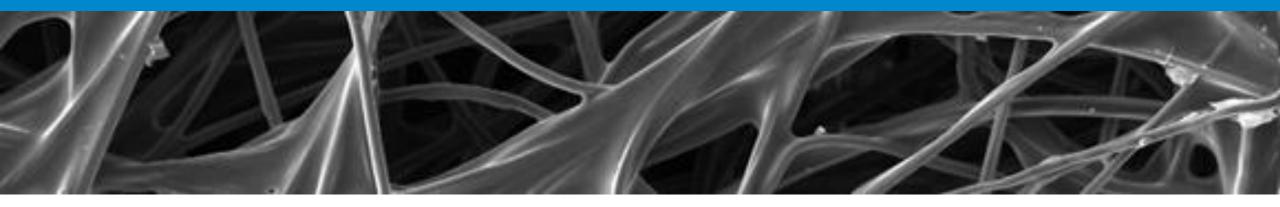
- Strength
- Softness / Stiffness
- Adhesion
- Firmness
- Durability
- Abrasion resistance
- Hydrophilicity / Hydrophobicity
- Reduced surface tension
- Improved dimension stability
- Solvent, wash & acid resistance
- ...

- Create added value products
- Achieve desired products requirements



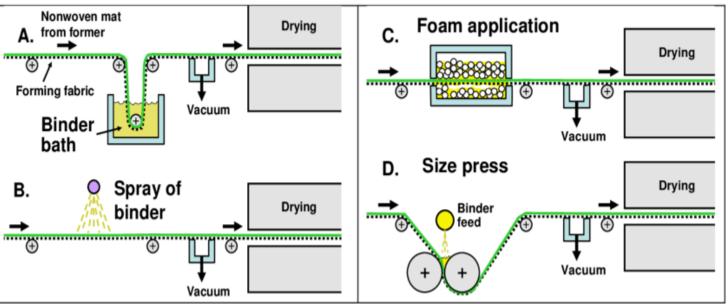
Types of chemical bonding





- Saturation
- Spray bonding
- Foam bonding
- Print bonding



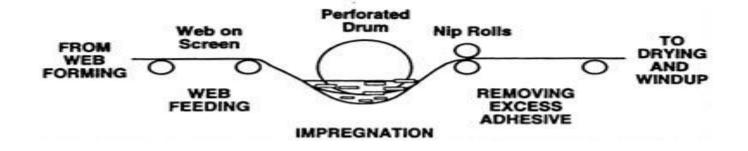




Saturation



- Complete immersion of the nonwoven web in a bath containing binder
- Excess binder is removed by vacuum or roll press
- For fabric application which require strength, stiffness and maximum fiber encapsulation
- Amount of binder uptake depends on
 - basis weight of nonwoven
 - time spent in bath
 - wettability of fiber
 - nip pressure

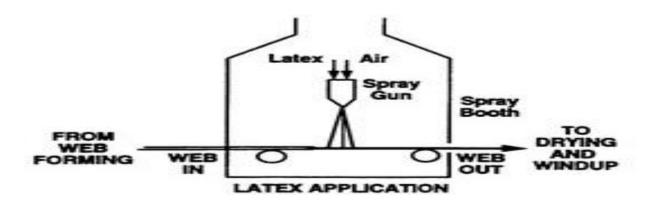




Spray bonding

4

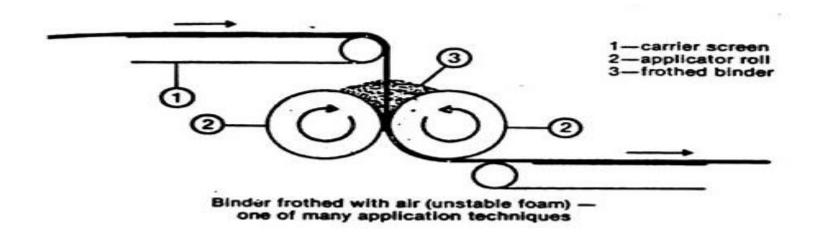
- Binder is sprayed onto moving web
- used for fabric applications which require the maintenance of high loft or bulk





Foam bonding

- Means to apply at low water and high binder-solids concentration levels
- Require less energy in drying, since less water is used compared to saturation

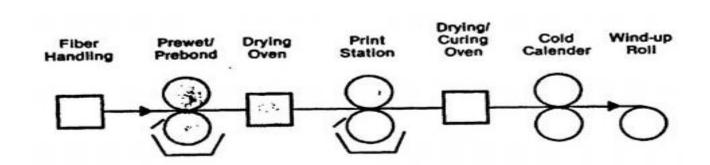






Print bonding

- Applies binder only in the predetermine areas
- Is used for fabric applications which require binder free area on the fabric





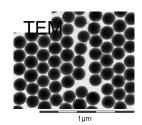
Definition of Latex and Dispersion / Emulsion

- Latex: liquid (Latin), droplet (Greek)
 - → naturally occurring rubber milk



- <u>Definition (Latex)</u>: A colloidal dispersion of polymer particles in an aqueous medium. The polymer may be organic or inorganic.
 - → aqueous synthetic organic polymer colloids, esp. for butadiene-styrene copolymer emulsions





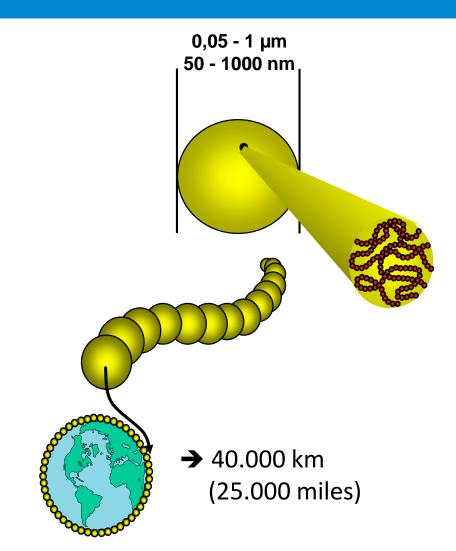
Definition (Polymer dispersion): A dispersion in which the disperse phase consists of polymer particles.





Latex in number and size

- Specification of "typical" Latex
 - Solid content 50%
 - Particel Ø 150nm
 - 1ml of dispersion contains 2,6 × 10¹⁴ particles
 - 1 particlecontains app. 10 10.000 macromolecular chains
 - 1 macromolecular chain contains app. 100 1.000.000 monomer units





Free radical chain polymerisation mechanisms



Initiation

■ decomposition of initiator I (formation of free radicals R•)

$$I \rightarrow 2R^{\bullet}$$

initiation of chain

$$R^{\bullet} + M \rightarrow RM_{1}^{\bullet}$$

$$R \star + CH_2 = CH \longrightarrow R - CH_2 - CH \star$$

Propagation

$$RM_{n}^{\bullet}$$
 + M \rightarrow RM_{n+1}^{\bullet}

$$R - CH_2 - CH^{\star} + n CH_2 = CH$$

$$X$$

$$X$$

$$R + CH_2 - CH + CH_2 - CH^{\star}$$

$$X$$

Termination

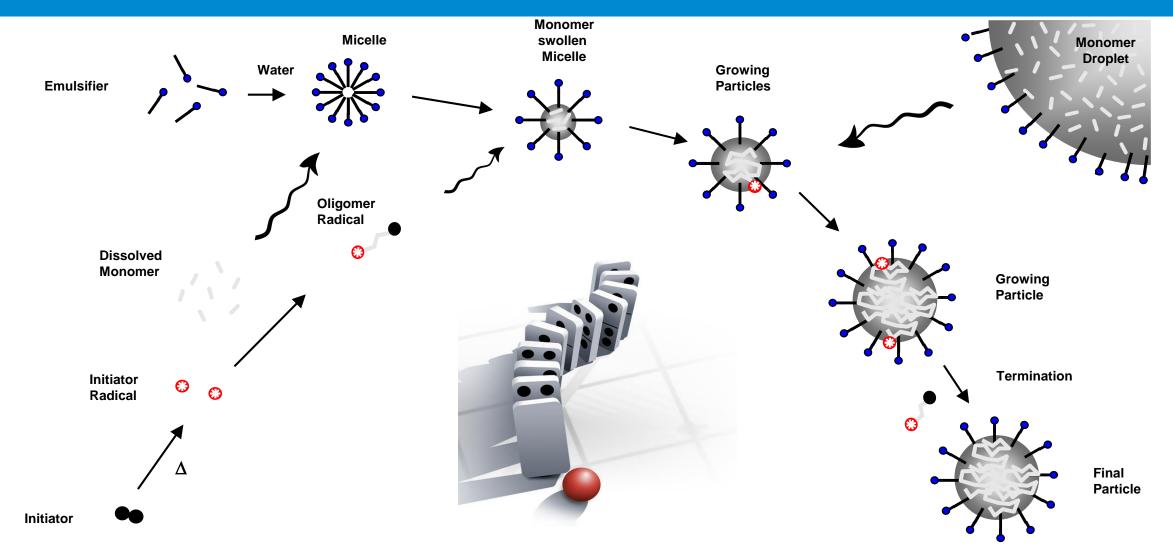
combination

Disproportionation



Emulsion polymerisation principle

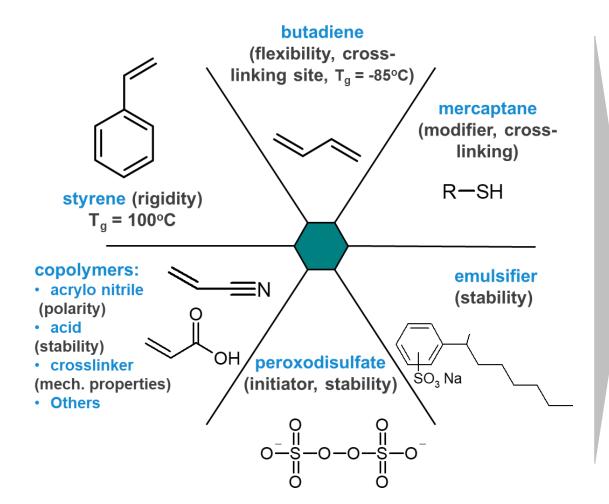


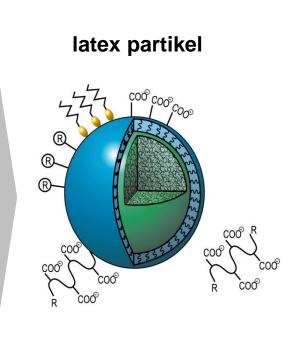




Main components for XSBR latices polymerisation











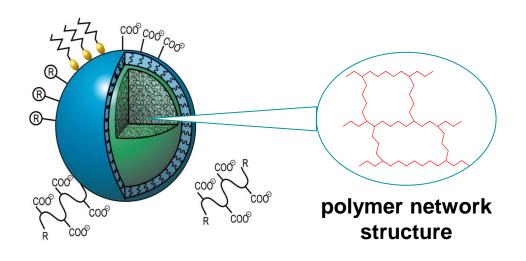
Crosslinking concept – building high polymer network



Concept of crosslinking

Creation of chemical bonds leads to increase of molecular weight and formation of polymer network

- I. Crosslinking during the polymer synthesis
 - monomers with 2 or more double bonds
 - Process conditions
- II.Crosslinking after polymer film formation
 - By functional group within the polymer backbone
 - More sophisticated the I.
 - Allows ideal wetting and bonding to substrate





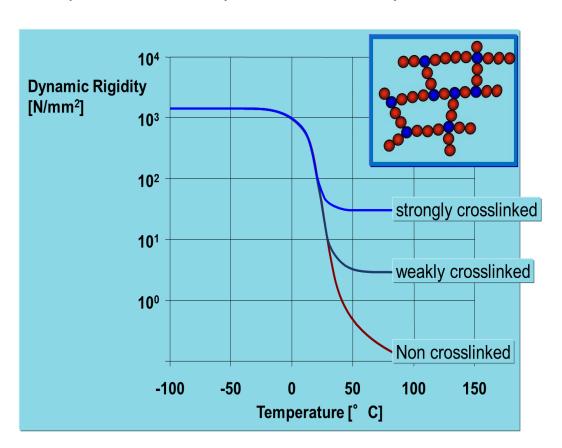
Crosslinking – Key for high performance Latices

Crosslinking leads to high performance on

- elasticity and resilience
- heat resistance
- tensile- / bonding strength
- resistance against solvents, acids, chemicals
- combination of soft handle & non blocking features
- better abrasion resistance

Crosslinking

Impact on elasticity and resilience by DMA



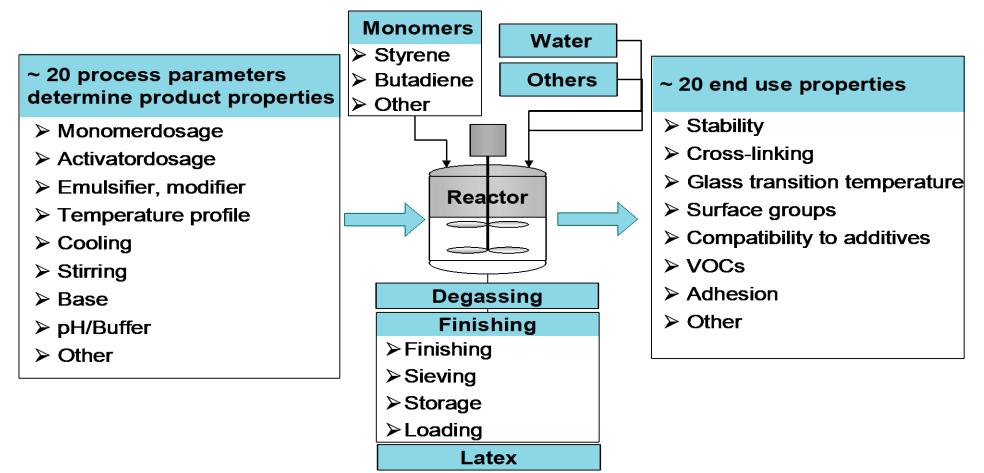


Emulsion polymerisation - Product development & process know-how



Emulsion polymerisation process is very flexible

billions of combinations of process parameter





4

Synthomers broad product portfolio



- xSBR carboxylated styrene-butadiene-rubber
- HS-SBR high-solid styrene-butadiene-rubber
- SA styrene acrylics
- PA pure acrylics
- NBR nitrile-butadiene-rubber
- VP-SBR vinyl-pyridine-styrene-butadiene-rubber
- CR chloroprene latex
- VAc Vinyl-acetate homopolymer

- VA-Co Vinyl-acetate co-polymers
- NR natural rubber

Apart from Synthomer portfolio

- EVA Ethylene-vinyl-acetate
- PUD Polyurethane-dispersion



xSBR - carboxylated styrene-butadiene-rubber

$$\begin{array}{c|c} - CH_2 - CH_2 - CH_2 - HC \\ \hline \end{array}$$

Key-Properties

- Hydrophobic
- Temperature resistant
- Alkaline resistant
- Good resilience
- Durability
- internal and external crosslink-able, vulcanizable
- Excellent combination flexibility and toughness

- Finishing of glass fibres (EIFS)
- Hygiene Nonwovens (ADL)
- PES Nonwoven for roofing
- Footwear
- Abrasives



HS-SBR – high-solid styrene-butadiene-rubber

Key-Properties

- Alternative for natural rubber
- very soft and elastic
- vulcanizable

- Mattresses
- Clutches
- footwear
- friction papers
- sealants



Pure acrylics & styrene acrylics



Key-Properties

- Excellent UV resistance
- Very good weathering resistance
- acid resistant
- solvent resistant
- internal and external crosslink-able

- Filter
- Tents
- Awnings
- Blinds
- Battery separators
- Technical paper
- Deco laminates



NBR - nitrile-butadiene-rubber

$$* - \begin{bmatrix} H_2 \\ C \end{bmatrix} - C = C - \begin{bmatrix} H_2 \\ C \end{bmatrix} - \begin{bmatrix}$$

Key-Properties

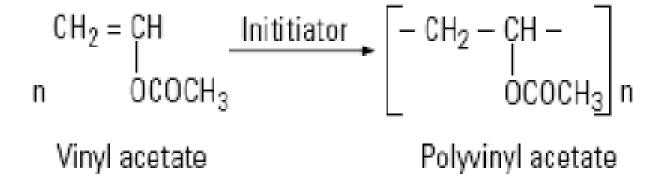
- Excellent Oil, fat & grease resistance
- chemical resistance
- wash resistance
- Very soft haptic, dry and tack free coatings
- internal and external crosslink-able, vulcanizable,
- adhesion promoter

- Sealants
- Gaskets
- Clutches
- Abrasives
- Artificial leather
- Wipes



VAc – Vinyl-acetate homopolymer & VA-Co – Vinyl-acetate co-polymers





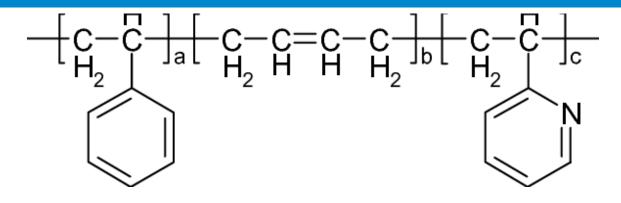
Key-Properties

- Hydrophobic
- Temperature resistant
- Alkaline resistant
- Good resilience
- Durability
- internal and external crosslink-able, vulcanizable
- Excellent combination flexibility and toughness

- Finishing of glass fibres mats
- Hygiene Nonwovens
- Footwear
- Abrasives



VP-SBR – vinyl-pyridine-styrene-butadiene-rubber



Key-Properties

- Excellent adhesion to Polyamide, PES, Aramide
- Hydrophobic
- Temperature resistant
- Durability
- internal and external crosslink-able, vulcanizable
- Excellent combination flexibility and toughness
- Adhesion promoter

- Adhesion promoter for rubber materials
- tire cord
- conveyor belts
- hoses



CR – chloroprene latex

$$-$$
CH₂ $-$ CH=CH $-$ CH₂ $-$ I

Key-Properties

- Excellent Oil, fat & grease resistance
- chemical resistance
- flame retardant
- wash resistance
- very soft
- vulcanizable

- Protection clothes
- Sealants
- Footwear





EVA – Ethylene-vinyl-acetate

Key-Properties

- High wet-strength
- Different hands possible
- More odor then other binders
- Excellent absorbency
- Cheap

- Wet wipes
- Air-laid pulp fabrics

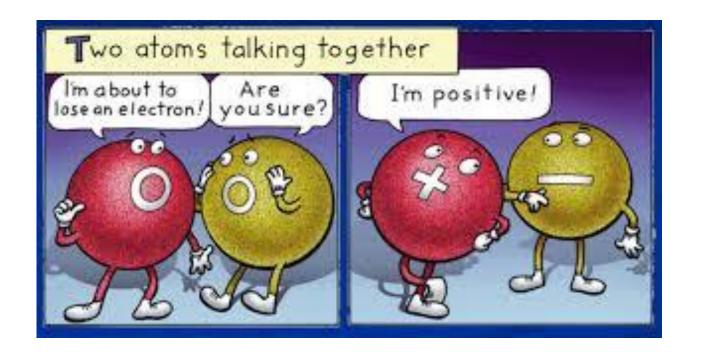


SUMMARY

Chemically bonded Nonwovens – adding value to the industry



- Different ways of web formation
- Different methods of web bonding
- Chemical bonding creates added value
- Emulsion polymerisation process very flexible
- Different chemistries deliver different properties
- → Synthomer as reliable & innovative supplier







HOW Can We Improve Your Nonwoven?



Matthias Renka November 16 - 17, 2019 34. Hofer Vliesstofftage, Hof - Germany

